



GRAYMONT

EFFECT OF LIME COAGULATION ON STRENGTH AND ATTERBERG PROPERTIES OF CLAYS IN OIL SANDS TAILINGS

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Outline

- ▶ Oil sands fine tailings
- ▶ Dewatering of oil sands fine tailings
- ▶ Lime as a coagulant of clay minerals
- ▶ Experimental results of lime as a coagulant of clay minerals in oil sands tailings
- ▶ Summary, conclusions, and future work

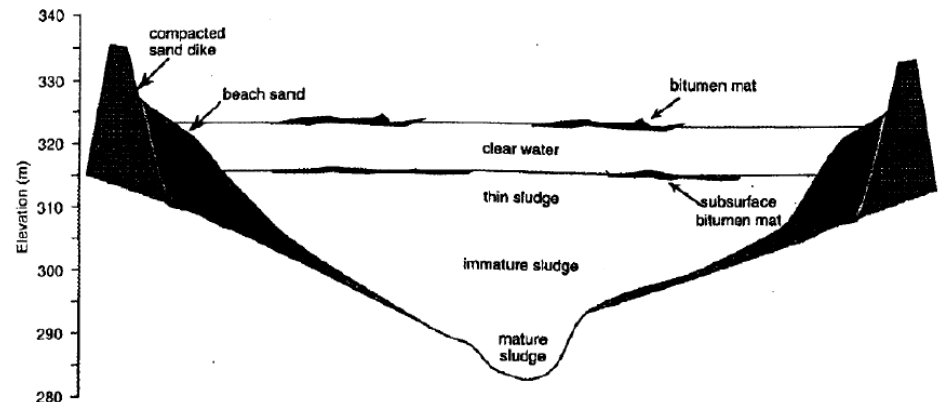
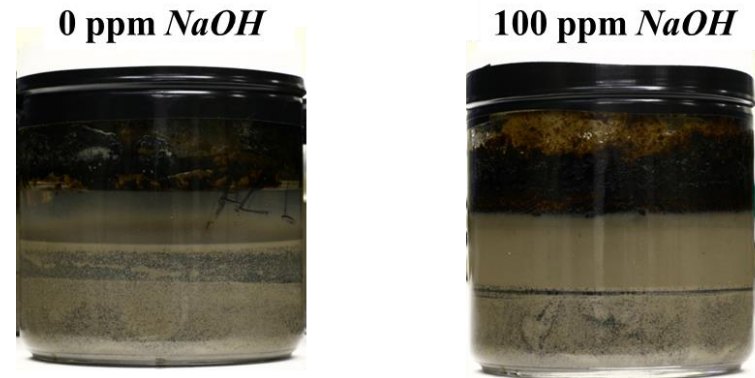
Oil sands fine tailings

► Extraction Process

- Ore-Water slurry extraction process
- Uses NaOH as a process additive to enhance recovery
- Extraction process results in highly dispersed clay minerals

► Tailings properties

- Tailings easily segregate upon beach deposition
- Fine silts and dispersed clay minerals collect in ponds as Fluid Fine Tailings (FFT)
- FFT settles slowly, resists consolidation, and is untrafficable



Mineralogy of oil sands fine tailings

► Most common clays in oil sands fine tailings are:

1. Kaolinite
2. Illite
3. Interstratified Smectites
4. Chlorite

Mineralogical composition of various fractions (± 10 wt.%)

Sample number	Fraction (μm)	Clay minerals (w/w)					Non-clay minerals (w/w)				
		Kaolin	K-S	Illite	I-S	Chlorite	Quartz	Anatase	Rutile	Siderite	Pyrite
MFT	≤ 0.25	15	37	2	46	—	—	—	—	—	—
	$\leq 2 > 0.25$	26	24	18	15	3	9	—	—	—	—
	$< 44 > 2$	20	—	19	—	5	55	—	1	—	—
Thickener overflow	#1	13	16	11	40	—	9	—	—	—	—
Thickener overflow	#2	7	24	3	55	—	7	—	—	—	—
Froth solids	≤ 2	25	29	5	29	—	7	2	0.4	—	3
	$\leq 10 > 2$	14	—	11	—	—	23	1	2	17	5
	$< 44 > 10$	7	—	12	—	0.7	45	2	2	13	12

Other minerals in the -44 - to $+10$ - μm fraction of the froth solids include calcite (2), dolomite (1), albite (1.2), K-feldspar (1) and magnetite (0.9).

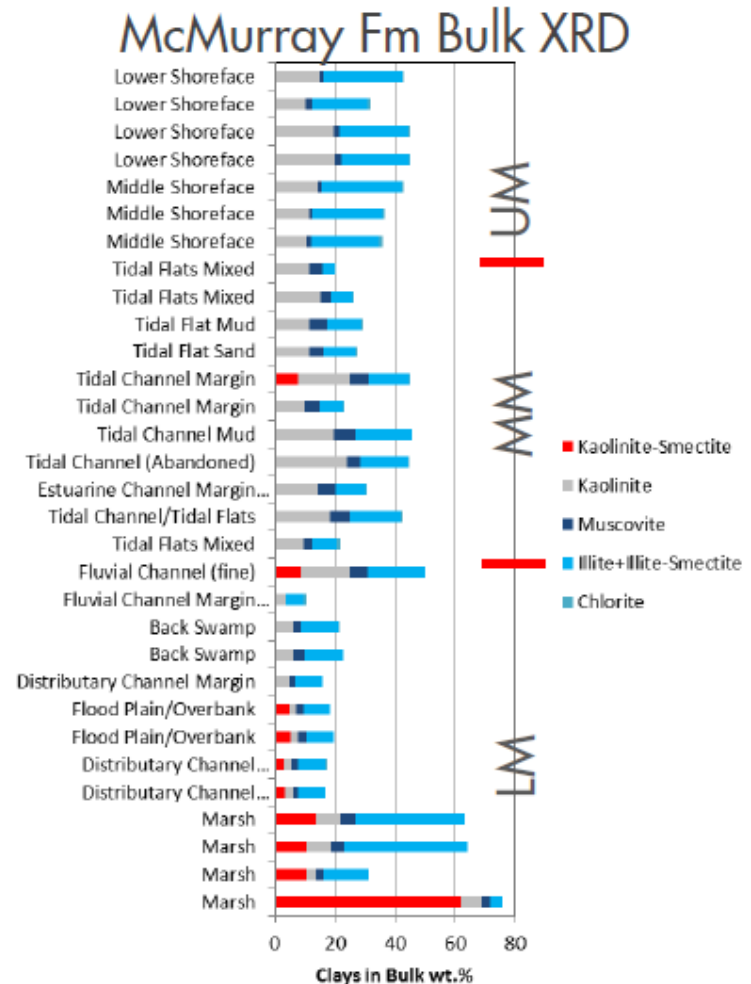
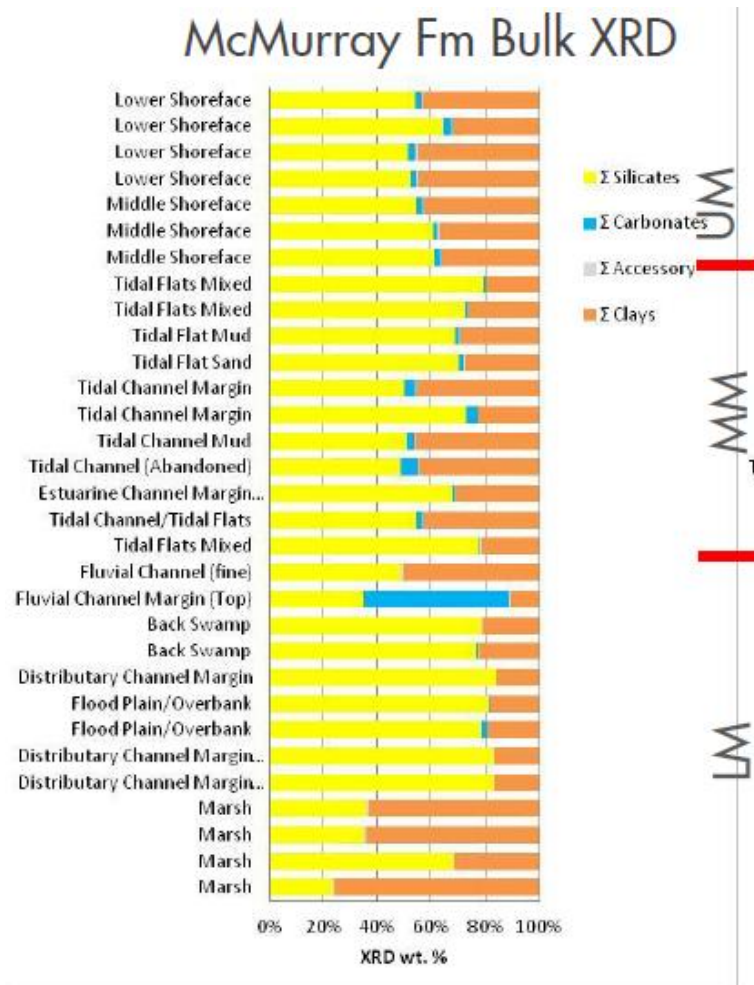
O.E. Omotoso, R.J. Mikula / Applied Clay Science 25 (2004) 37–47

► Abundance of clays is highly dependent on geological formations and facies

- Interstratified Smectites are most common in fluvial facies
- Illite is most common in marine facies

► All fine tailings will have different distributions of clays due to variability in ore bodies across operator leases

Mineralogy of oil sand tailings



Day-Stirrat, R, et al, 2015 Oil Sands Clay Conference

What is dewatering in oil sands fine tailings?

► Phases of dewatering

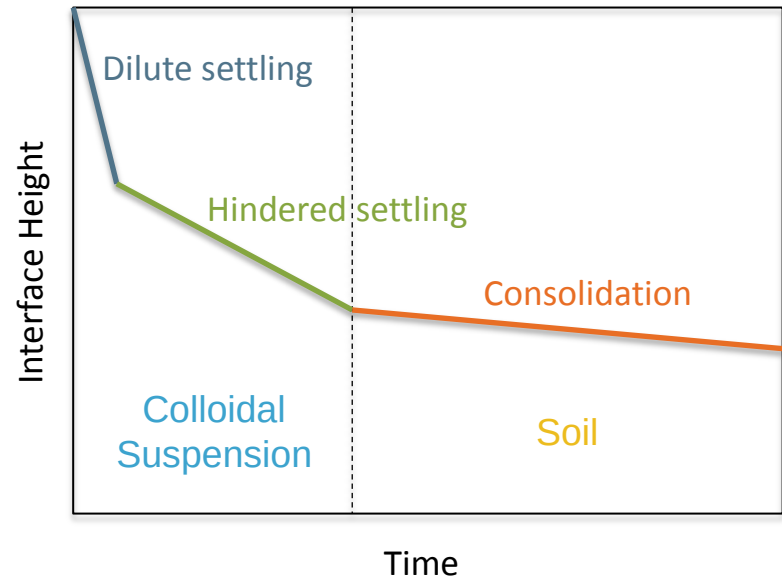
Colloidal Suspension

- **Dilute settling**
 - Initial quick settling
 - Limited particle interaction
 - Ex: In-line flocculation at 0-20% solids content
- **Hindered Settling**
 - Slow settling
 - Significant particle interaction
 - Ex: Post thickener at 20%+ solids content
- **Consolidation**
 - Particle interaction dominates
 - Generally occurs with external load
 - Ex: Overburden co-mixing at 70%+ solids content

Soil

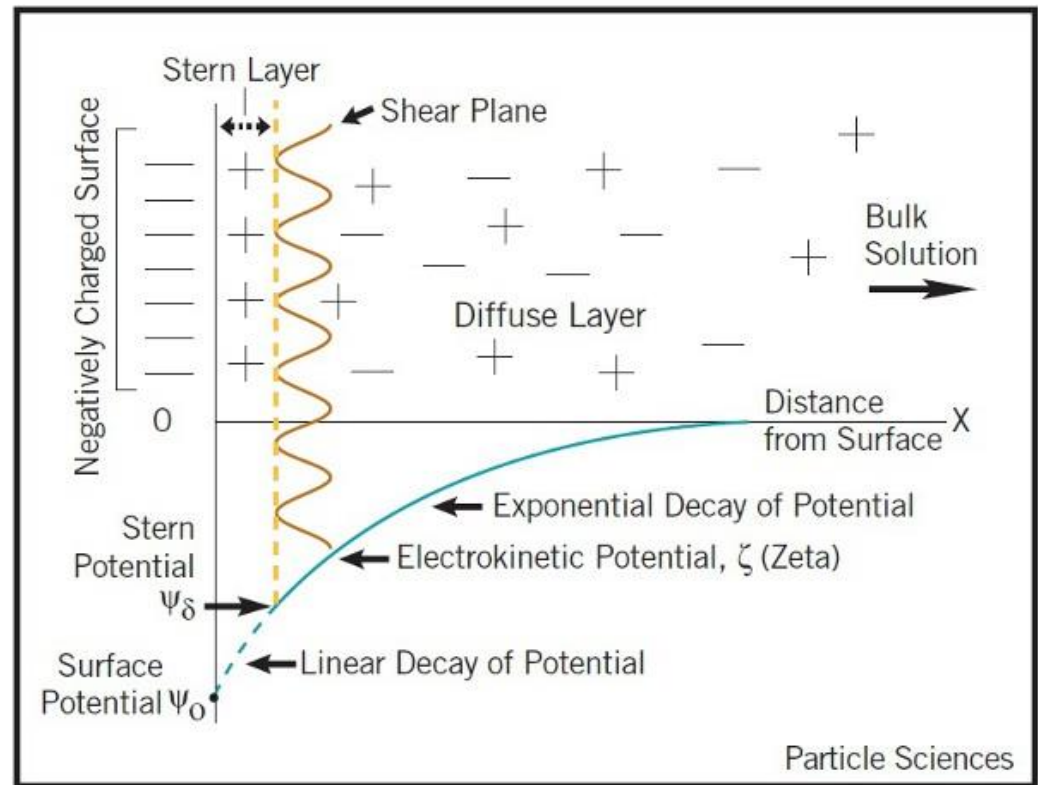
► Effect of mineral surfaces on dewatering

- Particle size and surface area
 - Coagulation vs Flocculation
- Water film thickness
 - Ability of clay minerals to swell with water
 - Controlled by cation exchange



Where is water in oil sand fine tailings?

- ▶ **Water is bound in the hydration shell of ions associated with clay surfaces**
- ▶ **Bulk water**
 - Ions have no attraction to surface
 - Easy to remove
- ▶ **Diffuse layer**
 - Ions have some attraction to surface
 - Beyond the shear plane, so removal is somewhat easy
- ▶ **Stern layer**
 - Strong attraction to charged surface
 - Removal is very difficult and requires modification of the surface, such as ion exchange

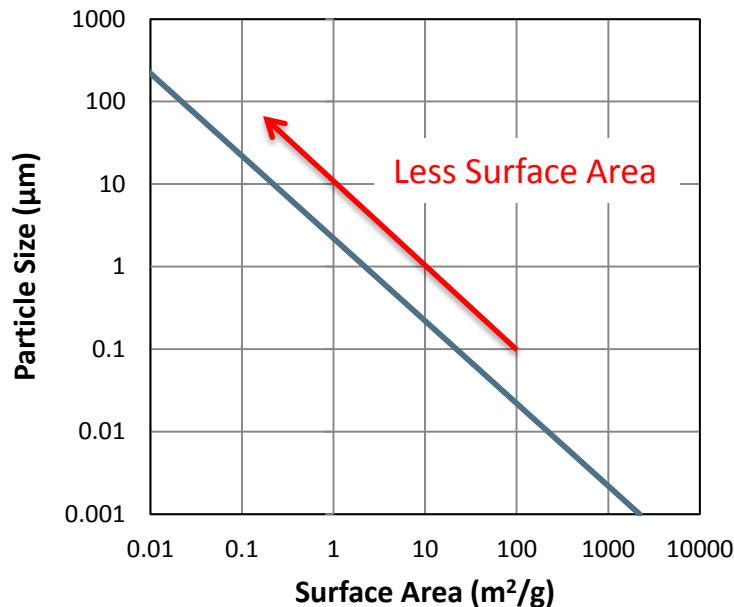


<http://media.americanpharmaceuticalreview.com/m/28/article/133232-2.jpg>

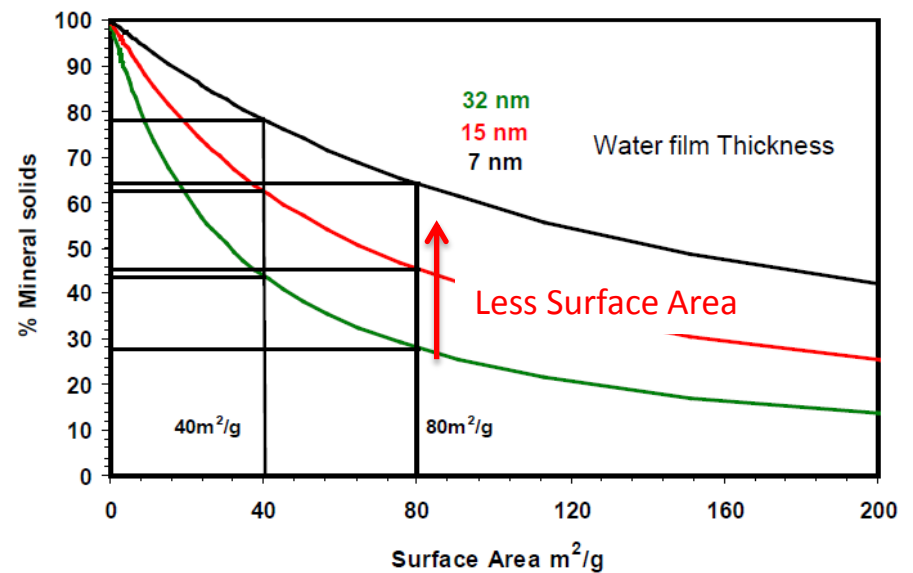
Changing surface properties can improve dewatering

- ▶ Decreasing clay surface area will provide more free water for removal
- ▶ Can be achieved by:

- Increasing clay particle size



- Changing surface chemistry to decrease water film thickness

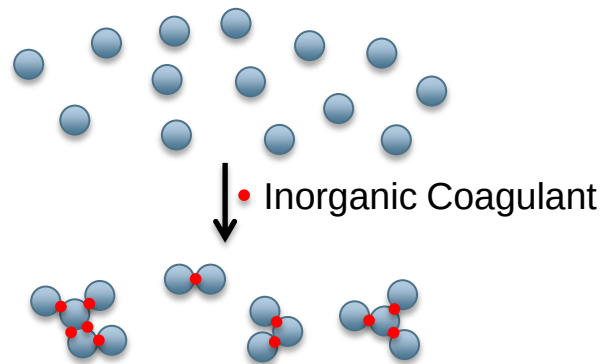


Sethi, A, 2015 Oil Sands Clay Conference

Coagulation vs flocculation in oil sands fine tailings

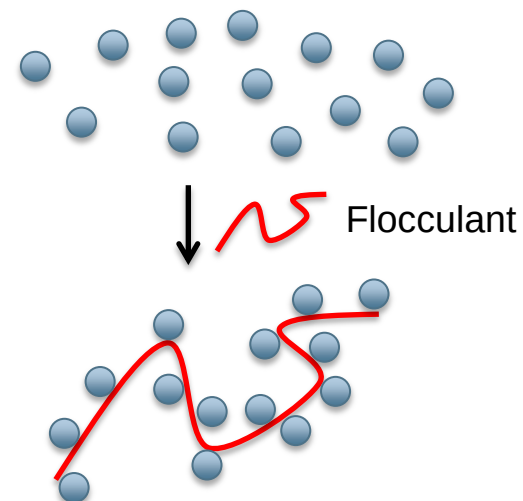
► Coagulation

- Aggregation of clay particles through chemical bonding
- Modification and chemical reaction of clay surfaces by coagulants, such as lime, by ion exchange and pozzolanic reactions
- Results in particles of larger size held together by chemical bonds of varying strengths
- Improves all dewatering through reduced surface area resulting from increased particle size and reduced water film thickness



► Flocculation

- Aggregation of clay particles onto a flocculant surface
- Particles adsorption onto large molecular weight flocculants, such as polyacrylamide polymers
- Results in very large and fragile flocs held together by weak intermolecular forces
- Enhances effects of gravity settling in dilute settling, accelerating initial dewatering



How much dewatering is needed in oil sands fine tailings?

- ▶ **Significant dewatering is currently necessary because oil sands tailings have low strengths and high plasticity, thus a high solids content is necessary to make a trafficable material**
 - Oil sands fine tailings need to be above about 70% solids to be a trafficable material
 - Current technologies have success dewatering to 50% solids; however, remaining water is much more difficult to remove as much is associated with the water film in the stern layer

- ▶ **Chemical modification of clays to increase strength and decrease plasticity of oil sands tailings could provide a trafficable material at a lower solids content**
 - Tailings that are stable and stackable at a lower solids content require less dewatering, reducing treatment and processing costs
 - Chemical modification of clay surfaces can decrease surface area and release free water which should improve dewatering processes

Lime as a coagulant of clay minerals

- ▶ **Lime, calcium oxide, provides calcium as Ca^{2+} , an inorganic coagulant**
- ▶ **Lime coagulation occurs by two mechanisms, ion exchange and pozzolanic reactions, dependent on pH**
 - In ion exchange, quick process where calcium bonds bridge clay particles
 - In pozzolanic reactions, slower cementitious stabilization resulting from the production of hydrated calcium silicates and calcium aluminates at high pH
- ▶ **Lime has significant impacts on process water chemistry**
 - Lime will initially react with bicarbonate and carbonates to form calcite
 - Lime will increase pH of system, which is necessary to form pozzolans
- ▶ **Lime can decrease plasticity and increase strength of clay materials**
 - Lime stabilization is well known to decrease plasticity index and increase shear strength
 - Provides a textural change from clay-like to a silt-like character

Investigation of lime coagulation in oil sands tailings

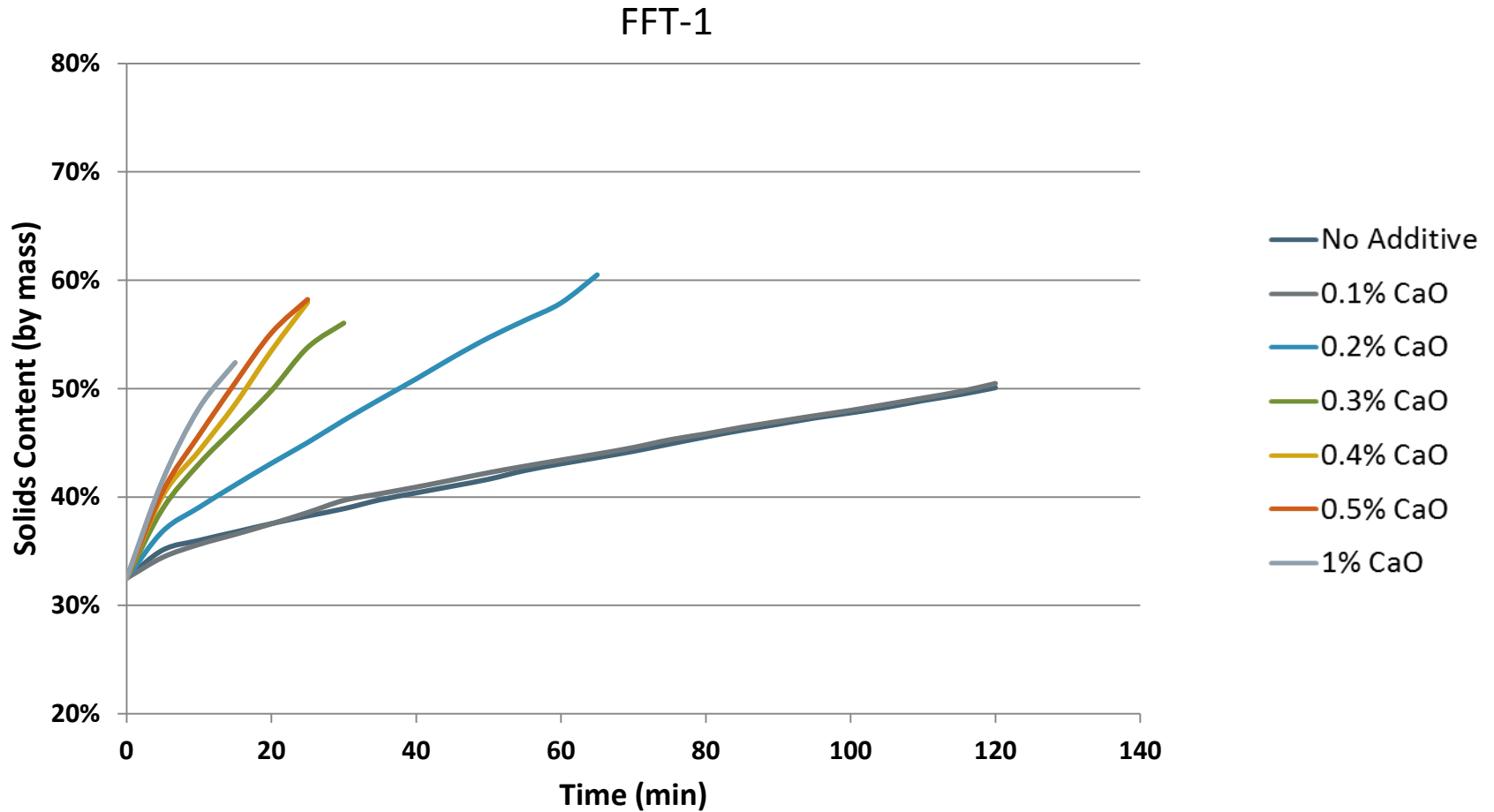
► **Fluid fine tailings (FFT) were treated with different doses (as a %Wt) of lime and investigated for:**

- Water release under pressure
- Particle size analysis
- Yield stress as a function of aging
- Atterberg limits

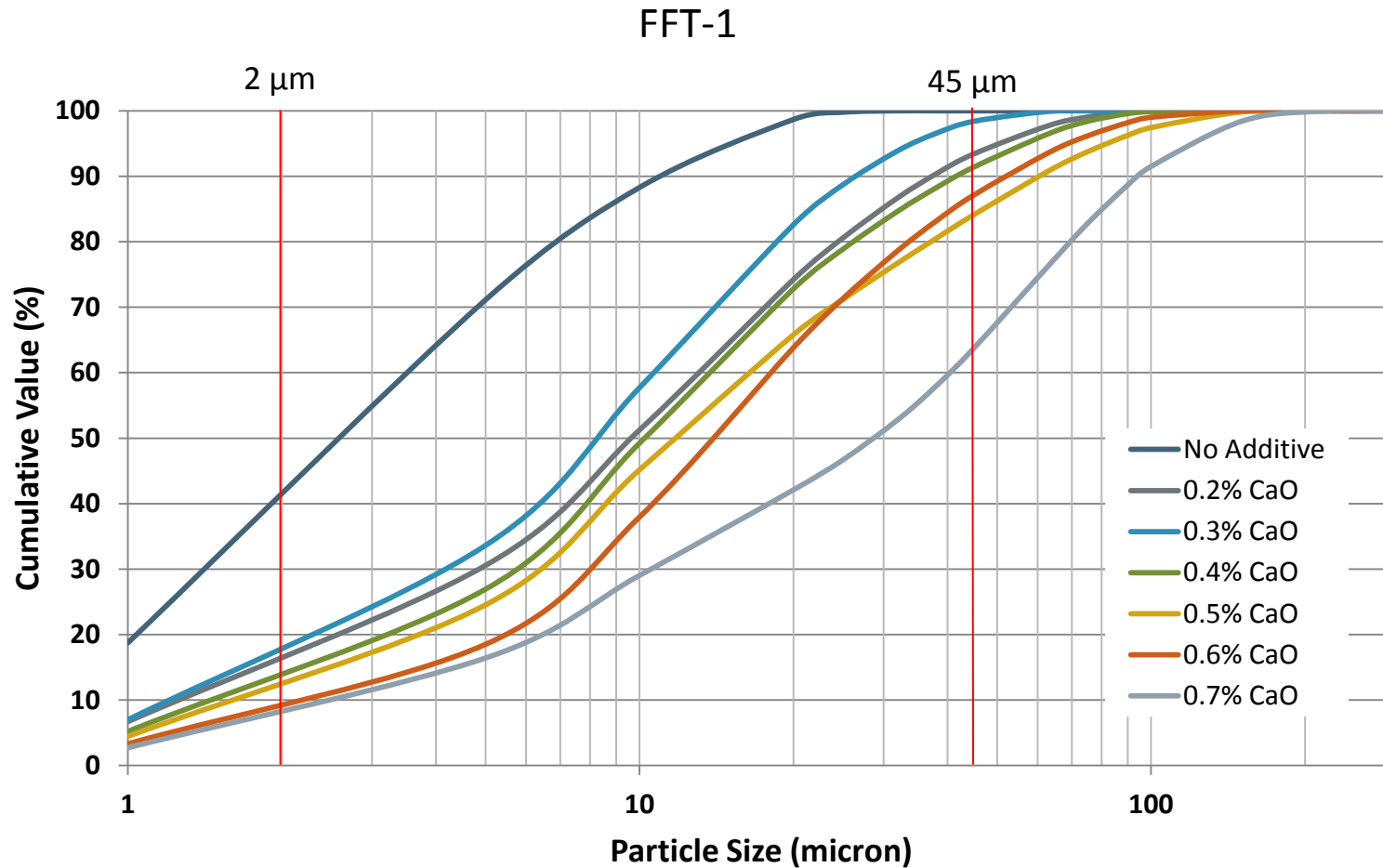
Characterization of FFT used in experiments

FFT	Mineral Solids Content	Bitumen Content	MBI	Na+	Carbonate Alkalinity	pH
	%	%		ppm	ppm	
1	32.5	2.9	9.2	362	579	7.9
2	33.4	1.3	11.2	877	937	8.3

Effect of lime coagulation for dewatering oil sands fine tailings under pressure

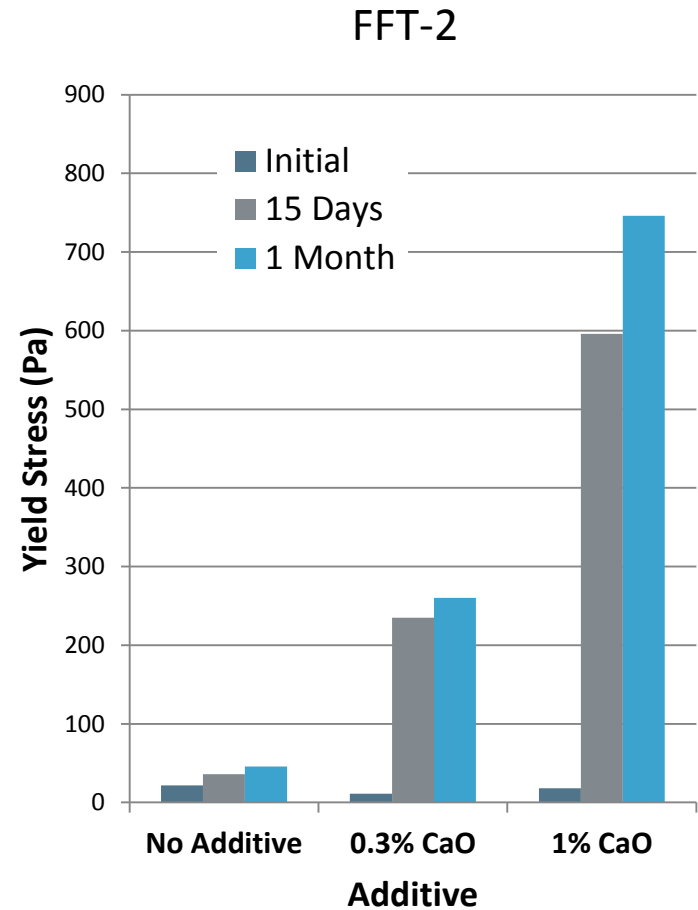
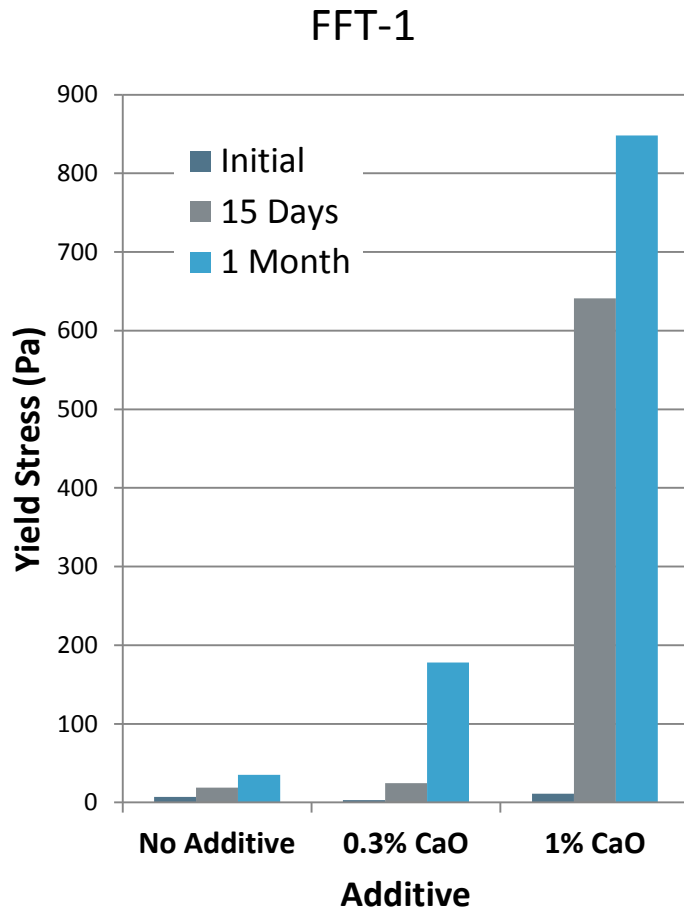


Effect of lime coagulation on oil sands tailings particle size

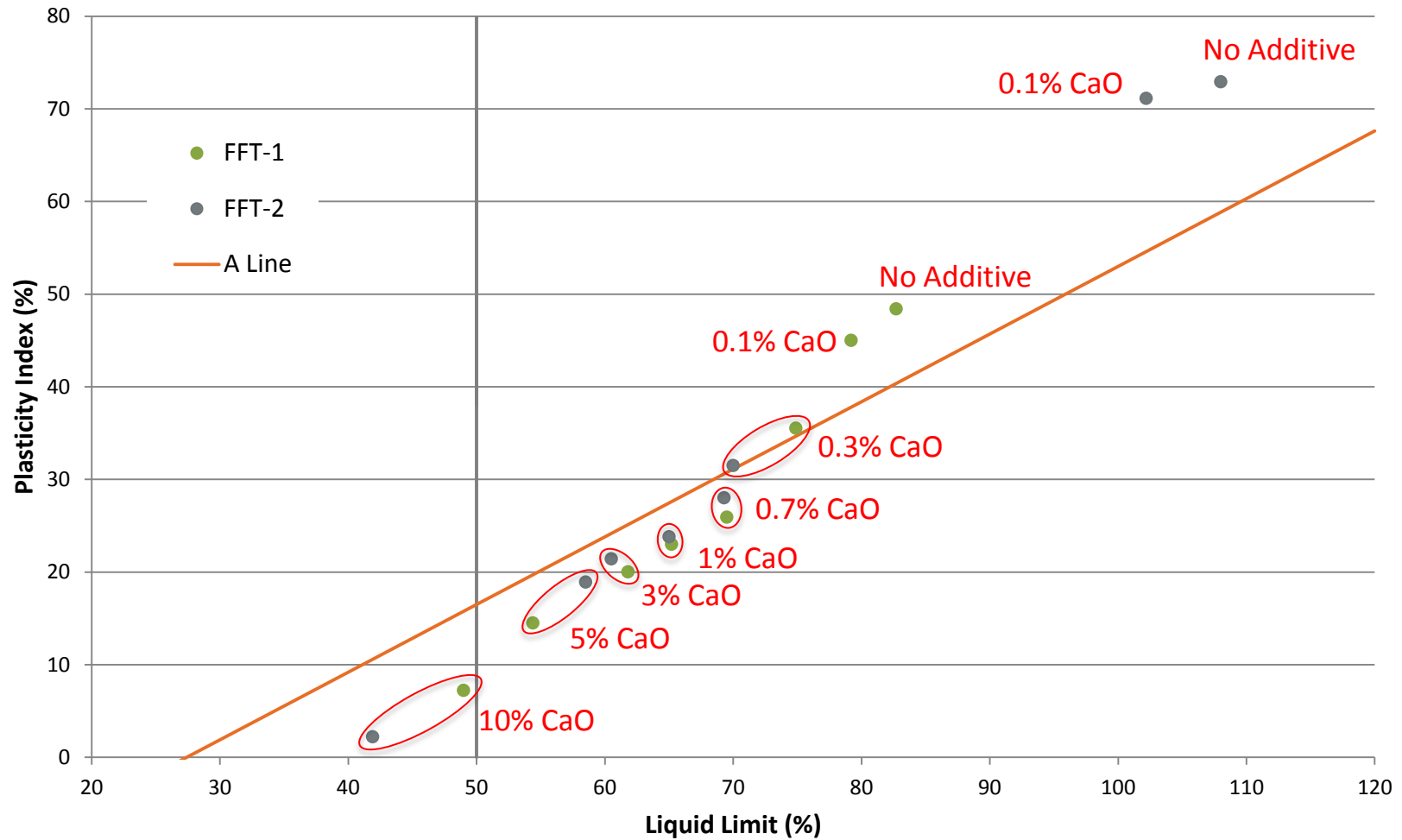


- pH above 12 was achieved with 0.7% Wt CaO dose

Effect of lime coagulation on oil sands fine tailings strength



Effect of lime coagulation on oil sands fine tailings plasticity



Summary of CaO Treated FFT Properties

► CaO treatment enhances dewatering under pressure

- Dose dependent response, enhanced dewatering under pressure was observed starting at 0.2% Wt CaO and was maximized between 0.4% to 1% Wt CaO
- Demonstrated an increase from 30 to 60% in solids content
- Rate of dewatering was enhanced significantly, with the rate increasing with dose

► CaO treatment is observed to increase the size of FFT particles

- Low doses provide a small increase in particle size
- A significant increase in size is observed when treatment conditions reach above pH 12 with 0.7% Wt CaO dose, potentially due to pozzolanic reactions

► CaO treatment increases strength as observed in yield stress measurements

- The 1% Wt CaO dose, which increases pH above 12, is observed to have a significant increase in the observed yield stress, increasing to about 700 Pa with 1% Wt compared to 200 Pa with 0.3% Wt and about 40 Pa with no treatment
- Strength appears to increase overtime

► CaO treatment is observed to significantly reduce both liquid limit and plasticity index

- CaO treatment appears to change FFT properties such that they appear to behave more silt-like
- Change to silt like character occurs at 0.7% Wt CaO, which corresponds to pH increasing above 12

Conclusions

- ▶ **Improvements to dewatering most likely results from a decrease in surface area of clays in oil sand fine tailings**
 - Surface area decreases by increase in particle size and chemical modification of clay surfaces which decreasing the water film thickness surrounding the clays
- ▶ **Improvement of strength and decrease in plasticity should allow for trafficable tailings at lower solids contents, thus less dewatering should be necessary to reclaim tailings**

Future work

- ▶ Investigation and characterization of pozzolanic reactions that occur in lime treated oil sands fine tailings
- ▶ Determination of undrained shear strength in lime treated oil sands fine tailings